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Chapter 11 - Thermochemistry Heat and Chemical Change Adapted from notes by Stephen Cotton 2 Section 11.1 The Flow of Energy - Heat XOBJECTIVES: • Explain the relationship between energy and heat. • Distinguish between heat capacity and specific heat. 3 Energy and Heat XThermochemistry - study of changes that occur during chemical reactions

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### SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below. a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity \_\_\_\_ 1. quantity of heat needed to raise the temperature of 1 g of water by 1°C \_\_\_\_ 2. SI unit of energy \_\_\_\_ 3.

### Thermochemistry Guided Reading Answers

To answer Questions 13 and 14, look at Figure 17.2 on page 506. 13. A system is a person sitting next to a campfire. Is this system endothermic or exothermic? Explain why. ... Thermochemistry (continued) GUIDED PRACTICE PROBLEM 12 (page 513) 12. ... Report "Guided Reading Key Ch17" Your name.

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THERMOCHEMISTRY SECTION 17.1 THE FLOW OF ENERGY-HEAT ... To answer Questions 13 and 14, look at Figure 17.2 on page 506. 13. A system is a person sitting next to a campfire. Is this system endothermic or ... 184 Guided Reading and Study Workbook @ Name \_ Date \_ Class \_ ~ Units for Measuring Heat Flow (page 507) 15.

### Chapter 11: Thermochemistry-Heat and Chemical Change

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### THERMOCHEMISTRY

Chapter 17 Thermochemistry. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. mbenjj0x0xx. Terms in this set (30) thermochemistry. study of energy changes that occur during chemical reactions and changes in state. chemical potential energy. energy stored in chemical bonds of a substance.

### Ch 17 Thermochemistry Practice Test

thermochemistry is the study of how energy in the form of heat is involved in chemical change. Kinetic and Potential Energy Energy takes many forms such as mechanical, electrical, or gravitational. These are categorized into two broad classes: kinetic energy, energy associated with motion, and potential energy, energy associated with position.

### Guided And Study Workbook Thermochemistry Answers

Thermochemistry Thermochemistry and Energy and Temperature Thermochemistry is study of changes in energy (heat) associated with physical or chemical changes. Force = push F= m a (mass x acceleration) force units: N (newton) = kg m s-2 Work = force x distance W = F d energy units: J (joule) = kg m2 s-2

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### Chapter 17 Thermochemistry Guided Reading And Study ...

184 Guided Reading and Study Workbook CHAPTER 17,Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the point of view of the system, or of the surroundings? 7. What universal law states that energy can neither be created nor destroyed

### Chapter 5: Thermochemistry

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