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Every minute of the
day, chemical
reactions take
place—both inside you
and around you. •After
a meal, a series of
chemical reactions.
take place as your

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body digests food.

- Plants use sunlight to drive the.

photosynthetic

processes needed to

produce. plant growth.

Section 2-4

Chemical Reactions and Enzymes

11.2 Types of Chemical

Reactions > 30 A

halogen can also

replace another

halogen from a

compound. •The

activity of halogens

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decreases as you go down Group 7A of the periodic table—fluorine, chlorine, bromine, and iodine. •Bromine is more active than iodine, so this reaction occurs: $\text{Br}_2(\text{aq}) + 2\text{NaI}(\text{aq}) \rightarrow 2\text{NaBr}(\text{aq}) + \text{I}_2(\text{aq})$

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Key Terms

Flashcards | Quizlet

1) All chemical reactions are chemical changes. 2) New properties appear. 3) The bonds between the atoms are broken and the atoms recombine 'n new ways. The carbon dioxide gas was produced through the chemical reaction in the bag. Be sure students don't misinterpret it as a

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Objectives
Vocabulary Part
ACompletion

Section 2-4 Chemical
Reactions and
Enzymes Date _____
(pages 49-53) Key
Concepts • What
happens to chemical
bonds during chemical

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reactions? • How do energy changes affect whether a chemical reaction will occur? • Why are enzymes important to living things? Chemical Reactions (page 49) 1. What is a chemical reaction?

Prentice Hall Chemistry Chapter 11: Chemical Reactions ...

a number in front of a chemical formula in an

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equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. synthesis. a chemical reaction in which two or more simple substances combine to form a new, more complex substance. decomposition.

Chapter 2

Prentice Hall Chemistry

Chapter 11: Chemical

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Reactions / Practice
Exam. Exam
Instructions: Choose
your answers to the
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'Next' to see the next
set of questions. You
can skip questions if
you would like and
come back to them
later with the yellow
"Go To First Skipped
Question" button.

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Key Terms and Definitions

reactions. A chemical reaction is a process that changes one set of chemicals into another set of chemicals. The elements or compounds that enter into the reaction are the reactants. The elements or compounds produced by the reaction are the products. Chemical reactions involve changes in the chemical bonds that join atoms in

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Bind

compounds. Energy in
Reactions Some
chemical reactions
release energy; others
absorb energy.

11.2 Types of Chemical Reactions>

A chemical reaction
can be concisely
represented by a
chemical 1.. The
substances that
undergo a chemical
change are the 2.. The
new substances
formed in a chemical

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Reactions Answer

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reaction are the 3.. In accordance with the law of conservation of ,
4. a chemical equation must be balanced.

When balancing an 5. equation, you place in front of reactants and products so 6.

05 CTR ch11 7/9/04

3:33 PM Page 265

DESCRIBING

CHEMICAL ...

d. reaction in which conversion of reactants to products and

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Chemical

Reactions Answer

Key Reaction Hill

Bind
products to reactants
occur simultaneously e.
ratio of product
concentrations to
reactant

concentrations with
each raised to a power
given by the number of
moles of the substance
in the balanced
equation Column A

_____ 16. reversible

reaction _____ 17.

chemical equilibrium

2.4 Chemical Reactions and

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Reactions Answer

Enzymes - Weebly

Section 2-4 Chemical

Reactions and

Enzymes(pages 49-53)

... reaction often

compared to a lock and

key?The active site and

the substrates have

complementary

shapes, and the fit is

very precise. 14. ...

Answer the questions

by writing the correct

vocabulary term in the

blanks.

LPT 2.4 - Chemical

Page 19/25

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Rx's & Enzymes.pdf -

Winston Knoll ...

2.4 Chemical Reactions

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IDEA Chemistry as the

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Chapter 11

64 For Practice

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Key, Purpose, Full

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direct students to correlating places in the eText and media. Empower students to visualize and understand chemistry. New Chapter 5 creates more topically-focused coverage—Chemical Reactions and Solutions have been divided into two separate chapters ...

Chapter 11 Chemical Reactions Test

Answer Key »

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chemical equation. a representation of the chemical reaction; the formulas of the reactants are connected by an arrow with the formulas of the products. skeleton equation. a chemical equation that does not indicate the relative amounts of the reactants and products.

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Reactions Answer

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g o < g
E .2 g SO gq6u

3uÅqqnd O . Z O v' u U

0 2 Mt E o u 0 u g a ...

Write correct formulas

for the products in

these decomposition

reactions. 3(3) a- (k)

coz (g) cs \checkmark C03 + COZ

8) 9) g 92RbC103 10)

(9 .

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The chemical reactions that occur inside the human body are extremely complex. One of the main reactions that occurs is the metabolism of glucose, and another reaction is the detoxification of the liver.

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